

Important Nuclear Magnetic Resonance Spectroscopy Formulas PDF



Formulas Examples with Units

List of 13 Important Nuclear Magnetic Resonance Spectroscopy Formulas

1) Chemical Shift in Nuclear Magnetic Resonance Spectroscopy Formula ↻

Formula

$$\delta = \left(\frac{\nu - \nu^\circ}{\nu^\circ} \right) \cdot 10^6$$

Example with Units

$$3\text{E}+8_{\text{ppm}} = \left(\frac{13_{\text{Hz}} - 10_{\text{Hz}}}{10_{\text{Hz}}} \right) \cdot 10^6$$

Evaluate Formula ↻

2) Effective Nuclear Charge given Shielding Constant Formula ↻

Formula

$$Z = z - \sigma$$

Example

$$17.5 = 18 - 0.5$$

Evaluate Formula ↻

3) Effective Transverse Relaxation Time Formula ↻

Formula

$$T2' = \frac{1}{\pi \cdot \Delta\nu_{1/2}}$$

Example with Units

$$21.2207_{\text{s}} = \frac{1}{3.1416 \cdot 0.0151_{\text{s}}}$$

Evaluate Formula ↻

4) Gyromagnetic Ratio given Larmor Frequency Formula ↻

Formula

$$\gamma = \frac{\nu_L \cdot 2 \cdot \pi}{(1 - \sigma) \cdot B_0}$$

Example with Units

$$5.236_{\text{C/kg}} = \frac{7.5_{\text{Hz}} \cdot 2 \cdot 3.1416}{(1 - 0.5) \cdot 18_{\text{T}}}$$

Evaluate Formula ↻

5) Hyperfine Splitting Constant Formula ↻

Formula

$$a = Q \cdot \rho$$

Example

$$6.3 = 2.1 \cdot 3$$

Evaluate Formula ↻

6) Local Distribution to Shielding Constant Formula ↻

Formula

$$\sigma_{\text{local}} = \sigma_{\text{d}} + \sigma_{\text{p}}$$

Example

$$27.1 = 7 + 20.1$$

Evaluate Formula ↻



7) Magnetogyric Ratio of Electron Formula

Formula

$$\gamma_e = \frac{e}{2 \cdot [\text{Mass-e}]}$$

Example with Units

$$8.8\text{E}+10 \text{ C/kg} = \frac{1.60\text{e-}19 \text{ C}}{2 \cdot 9.1\text{E-}31 \text{ kg}}$$

Evaluate Formula 

8) Nuclear Larmor Frequency Formula

Formula

$$\nu_L = \frac{\gamma \cdot B_{\text{loc}}}{2 \cdot \pi}$$

Example with Units

$$30.5577 \text{ Hz} = \frac{12 \text{ C/kg} \cdot 16 \text{ T}}{2 \cdot 3.1416}$$

Evaluate Formula 

9) Nuclear Larmor Frequency given Shielding Constant Formula

Formula

$$\nu_L = (1 - \sigma) \cdot \left(\frac{\gamma \cdot B_0}{2 \cdot \pi} \right)$$

Example with Units

$$17.1887 \text{ Hz} = (1 - 0.5) \cdot \left(\frac{12 \text{ C/kg} \cdot 18 \text{ T}}{2 \cdot 3.1416} \right)$$

Evaluate Formula 

10) Observed Width at Half-Height of NMR Line Formula

Formula

$$\Delta\nu_{1/2} = \frac{1}{\pi \cdot T_2}$$

Example with Units

$$0.0152 \text{ 1/s} = \frac{1}{3.1416 \cdot 21 \text{ s}}$$

Evaluate Formula 

11) Rate of Exchange at Coalescence Temperature Formula

Formula

$$k_c = \frac{\pi \cdot \Delta\nu}{\sqrt{2}}$$

Example with Units

$$35.5431 \text{ 1/s} = \frac{3.1416 \cdot 16 \text{ Hz}}{\sqrt{2}}$$

Evaluate Formula 

12) Shielding Constant given Effective Nuclear Charge Formula

Formula

$$\sigma = z - Z$$

Example

$$3 = 18 - 15$$

Evaluate Formula 

13) Total Local Magnetic Field Formula

Formula

$$B_{\text{loc}} = (1 - \sigma) \cdot B_0$$

Example with Units

$$9 \text{ T} = (1 - 0.5) \cdot 18 \text{ T}$$

Evaluate Formula 



Variables used in list of Nuclear Magnetic Resonance Spectroscopy Formulas above

- **a** Hyperfine Splitting Constant
- **B₀** Magnitude of Magnetic Field in Z-Direction (Tesla)
- **B_{loc}** Local Magnetic Field (Tesla)
- **e** Charge of Electron (Coulomb)
- **k_c** Rate of Exchange (1 Per Second)
- **Q** Empirical Constant in NMR
- **T₂** Transverse Relaxation Time (Second)
- **T2'** Effective Transverse Relaxation Time (Second)
- **z** Atomic Number
- **Z** Effective Nuclear Charge
- **γ** Gyromagnetic Ratio (Coulomb per Kilogram)
- **γ_e** Magnetogyric Ratio (Coulomb per Kilogram)
- **δ** Chemical Shift (Parts per Million)
- **Δv** Peak Separation (Hertz)
- **Δv_{1/2}** Observed Width at Half-Height (1 per Second)
- **v** Resonance Frequency (Hertz)
- **v_L** Nuclear Larmor Frequency (Hertz)
- **v°** Resonance Frequency of Standard Reference (Hertz)
- **ρ** Spin Density
- **σ** Shielding Constant in NMR
- **σ_d** Diamagnetic Contribution
- **σ_{local}** Local Contribution
- **σ_p** Paramagnetic Contribution

Constants, Functions, Measurements used in list of Nuclear Magnetic Resonance Spectroscopy Formulas above

- **constant(s):** pi, 3.14159265358979323846264338327950288
Archimedes' constant
- **constant(s):** [Mass-e], 9.10938356E-31
Mass of electron
- **Functions:** sqrt, sqrt(Number)
A square root function is a function that takes a non-negative number as an input and returns the square root of the given input number.
- **Measurement: Time** in Second (s)
Time Unit Conversion 
- **Measurement: Electric Charge** in Coulomb (C)
Electric Charge Unit Conversion 
- **Measurement: Frequency** in Hertz (Hz)
Frequency Unit Conversion 
- **Measurement: Magnetic Field** in Tesla (T)
Magnetic Field Unit Conversion 
- **Measurement: Radiation Exposure** in Coulomb per Kilogram (C/kg)
Radiation Exposure Unit Conversion 
- **Measurement: Salinity** in Parts per Million (ppm)
Salinity Unit Conversion 
- **Measurement: Vorticity** in 1 per Second (1/s)
Vorticity Unit Conversion 
- **Measurement: Time Inverse** in 1 Per Second (1/s)
Time Inverse Unit Conversion 



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