

Important Air-Standard Cycles Formulas PDF



Formulas Examples with Units

List of 18 Important Air-Standard Cycles Formulas

1) Actual Air Fuel Ratio Formula

Formula

$$R_a = \frac{m_a}{m_f}$$

Example with Units

$$15.9936 = \frac{23.9904 \text{ kg}}{1.5 \text{ kg}}$$

Evaluate Formula 

2) Air Standard Efficiency for Diesel Engines Formula

Formula

$$\eta_d = 100 \cdot \left(1 - \frac{1}{r^{\gamma-1}} \cdot \frac{r_c^{\gamma} - 1}{\gamma \cdot (r_c - 1)} \right)$$

Evaluate Formula 

Example

$$64.9039 = 100 \cdot \left(1 - \frac{1}{20^{1.4-1}} \cdot \frac{1.95^{1.4} - 1}{1.4 \cdot (1.95 - 1)} \right)$$

3) Air Standard Efficiency for Petrol engines Formula

Formula

$$\eta_o = 100 \cdot \left(1 - \frac{1}{r^{\gamma-1}} \right)$$

Example

$$69.8291 = 100 \cdot \left(1 - \frac{1}{20^{1.4-1}} \right)$$

Evaluate Formula 

4) Air Standard Efficiency given Relative Efficiency Formula

Formula

$$\eta = \frac{\eta_i}{\eta_r}$$

Example

$$0.506 = \frac{42}{83}$$

Evaluate Formula 



5) Mean Effective Pressure in Diesel Cycle Formula ↻

Formula

$$P_D = P_1 \cdot \frac{\gamma \cdot r^\gamma \cdot (r_c - 1) - r \cdot (r_c^\gamma - 1)}{(\gamma - 1) \cdot (r - 1)}$$

Evaluate Formula ↻

Example with Units

$$828.2159 \text{ kPa} = 110 \text{ kPa} \cdot \frac{1.4 \cdot 20^{1.4} \cdot (1.95 - 1) - 20 \cdot (1.95^{1.4} - 1)}{(1.4 - 1) \cdot (20 - 1)}$$

6) Mean Effective Pressure in Dual Cycle Formula ↻

Formula

$$P_d = P_1 \cdot \frac{r^\gamma \cdot ((R_p - 1) + \gamma \cdot R_p \cdot (r_c - 1)) - r \cdot (R_p \cdot r_c^\gamma - 1)}{(\gamma - 1) \cdot (r - 1)}$$

Evaluate Formula ↻

Example with Units

$$4348.961 \text{ kPa} = 110 \text{ kPa} \cdot \frac{20^{1.4} \cdot ((3.35 - 1) + 1.4 \cdot 3.35 \cdot (1.95 - 1)) - 20 \cdot (3.35 \cdot 1.95^{1.4} - 1)}{(1.4 - 1) \cdot (20 - 1)}$$

7) Mean Effective Pressure in Otto Cycle Formula ↻

Formula

$$P_O = P_1 \cdot r \cdot \left(\frac{(r^{\gamma-1} - 1) \cdot (r_p - 1)}{(r - 1) \cdot (\gamma - 1)} \right)$$

Evaluate Formula ↻

Example with Units

$$1567.7381 \text{ kPa} = 110 \text{ kPa} \cdot 20 \cdot \left(\frac{(20^{1.4-1} - 1) \cdot (3.34 - 1)}{(20 - 1) \cdot (1.4 - 1)} \right)$$

8) Relative Air-Fuel Ratio Formula ↻

Formula

$$\Phi = \frac{R_a}{R_i}$$

Example

$$1.088 = \frac{15.9936}{14.7}$$

Evaluate Formula ↻

9) Thermal Efficiency of Atkinson Cycle Formula ↻

Formula

$$\eta_a = 100 \cdot \left(1 - \gamma \cdot \left(\frac{e - r}{e^\gamma - r^\gamma} \right) \right)$$

Example

$$62.2417 = 100 \cdot \left(1 - 1.4 \cdot \left(\frac{4 - 20}{4^{1.4} - 20^{1.4}} \right) \right)$$

Evaluate Formula ↻



10) Thermal Efficiency of Diesel Cycle Formula

Formula

$$\eta_{th} = 1 - \frac{1}{r^{\gamma-1}} \cdot \frac{r_c^{\gamma} - 1}{\gamma \cdot (r_c - 1)}$$

Example

$$0.649 = 1 - \frac{1}{20^{1.4-1}} \cdot \frac{1.95^{1.4} - 1}{1.4 \cdot (1.95 - 1)}$$

Evaluate Formula 

11) Thermal Efficiency of Dual Cycle Formula

Formula

$$\varepsilon_d = 100 \cdot \left(1 - \frac{1}{r^{\gamma-1}} \cdot \left(\frac{R_p \cdot r_c^{\gamma} - 1}{R_p - 1 + R_p \cdot \gamma \cdot (r_c - 1)} \right) \right)$$

Example

$$66.6046 = 100 \cdot \left(1 - \frac{1}{20^{1.4-1}} \cdot \left(\frac{3.35 \cdot 1.95^{1.4} - 1}{3.35 - 1 + 3.35 \cdot 1.4 \cdot (1.95 - 1)} \right) \right)$$

Evaluate Formula 

12) Thermal Efficiency of Ericsson Cycle Formula

Formula

$$\eta_e = \frac{T_H - T_L}{T_H}$$

Example with Units

$$0.52 = \frac{250\text{K} - 120\text{K}}{250\text{K}}$$

Evaluate Formula 

13) Thermal Efficiency of Lenoir Cycle Formula

Formula

$$\eta_l = 100 \cdot \left(1 - \gamma \cdot \left(\frac{r_p^{\frac{1}{\gamma}} - 1}{r_p - 1} \right) \right)$$

Example

$$18.2442 = 100 \cdot \left(1 - 1.4 \cdot \left(\frac{3.34^{\frac{1}{1.4}} - 1}{3.34 - 1} \right) \right)$$

Evaluate Formula 

14) Thermal Efficiency of Otto Cycle Formula

Formula

$$\varepsilon_o = 1 - \frac{1}{r^{\gamma-1}}$$

Example

$$0.6983 = 1 - \frac{1}{20^{1.4-1}}$$

Evaluate Formula 



15) Thermal Efficiency of Stirling Cycle given Heat Exchanger Effectiveness Formula

Formula

Evaluate Formula 

$$\eta_s = 100 \cdot \left(\frac{[R] \cdot \ln(r) \cdot (T_f - T_i)}{[R] \cdot T_f \cdot \ln(r) + C_v \cdot (1 - \varepsilon) \cdot (T_f - T_i)} \right)$$

Example with Units

$$19.8854 = 100 \cdot \left(\frac{8.3145 \cdot \ln(20) \cdot (423\text{ K} - 283\text{ K})}{8.3145 \cdot 423\text{ K} \cdot \ln(20) + 100\text{ J/K}^{\circ}\text{mol} \cdot (1 - 0.5) \cdot (423\text{ K} - 283\text{ K})} \right)$$

16) Work Output for Diesel Cycle Formula

Formula

Evaluate Formula 

$$W_d = P_1 \cdot V_1 \cdot \frac{r^{\gamma-1} \cdot (\gamma \cdot (r_c - 1) - r^{1-\gamma} \cdot (r_c^{\gamma} - 1))}{\gamma - 1}$$

Example with Units

$$511.4233\text{ kJ} = 110\text{ kPa} \cdot 0.65\text{ m}^3 \cdot \frac{20^{1.4-1} \cdot (1.4 \cdot (1.95 - 1) - 20^{1-1.4} \cdot (1.95^{1.4} - 1))}{1.4 - 1}$$

17) Work Output for Dual Cycle Formula

Formula

Evaluate Formula 

$$W_D = P_1 \cdot V_1 \cdot \frac{r^{\gamma-1} \cdot (\gamma \cdot r_p \cdot (r_c - 1) + (r_p - 1)) - (r_p \cdot r_c^{\gamma} - 1)}{\gamma - 1}$$

Example with Units

$$2676.2321\text{ kJ} = 110\text{ kPa} \cdot 0.65\text{ m}^3 \cdot \frac{20^{1.4-1} \cdot (1.4 \cdot 3.34 \cdot (1.95 - 1) + (3.34 - 1)) - (3.34 \cdot 1.95^{1.4} - 1)}{1.4 - 1}$$

18) Work Output for Otto Cycle Formula

Formula

Evaluate Formula 

$$W_o = P_1 \cdot V_1 \cdot \frac{(r_p - 1) \cdot (r^{\gamma-1} - 1)}{\gamma - 1}$$

Example with Units

$$968.0783\text{ kJ} = 110\text{ kPa} \cdot 0.65\text{ m}^3 \cdot \frac{(3.34 - 1) \cdot (20^{1.4-1} - 1)}{1.4 - 1}$$



Variables used in list of Air-Standard Cycles Formulas above

- **C_v** Molar Specific Heat Capacity at Constant Volume (*Joule Per Kelvin Per Mole*)
- **e** Expansion Ratio
- **m_a** Mass of Air (*Kilogram*)
- **m_f** Mass of Fuel (*Kilogram*)
- **P₁** Pressure at Start of Isentropic Compression (*Kilopascal*)
- **P_d** Mean Effective Pressure of Dual Cycle (*Kilopascal*)
- **P_D** Mean Effective Pressure of Diesel Cycle (*Kilopascal*)
- **P_O** Mean Effective Pressure of Otto Cycle (*Kilopascal*)
- **r** Compression Ratio
- **R_a** Actual Air Fuel Ratio
- **r_c** Cut-off Ratio
- **R_f** Stoichiometric Air Fuel Ratio
- **r_p** Pressure Ratio
- **R_p** Pressure Ratio in Dual Cycle
- **T_f** Final Temperature (*Kelvin*)
- **T_H** Higher Temperature (*Kelvin*)
- **T_i** Initial Temperature (*Kelvin*)
- **T_L** Lower Temperature (*Kelvin*)
- **V₁** Volume at Start of Isentropic Compression (*Cubic Meter*)
- **W_d** Work Output of Diesel Cycle (*Kilojoule*)
- **W_D** Work Output of Dual Cycle (*Kilojoule*)
- **W_O** Work Output of Otto Cycle (*Kilojoule*)
- **γ** Heat Capacity Ratio
- **ε** Effectiveness of Heat Exchanger
- **ε_d** Thermal Efficiency of Dual Cycle
- **ε_O** Thermal Efficiency of Otto Cycle
- **η** Efficiency

Constants, Functions, Measurements used in list of Air-Standard Cycles Formulas above

- **constant(s):** **[R]**, 8.31446261815324
Universal gas constant
- **Functions:** **ln**, ln(Number)
The natural logarithm, also known as the logarithm to the base e, is the inverse function of the natural exponential function.
- **Measurement: Weight** in Kilogram (kg)
Weight Unit Conversion ↻
- **Measurement: Temperature** in Kelvin (K)
Temperature Unit Conversion ↻
- **Measurement: Volume** in Cubic Meter (m³)
Volume Unit Conversion ↻
- **Measurement: Pressure** in Kilopascal (kPa)
Pressure Unit Conversion ↻
- **Measurement: Energy** in Kilojoule (KJ)
Energy Unit Conversion ↻
- **Measurement: Molar Specific Heat Capacity at Constant Volume** in Joule Per Kelvin Per Mole (J/K*mol)
Molar Specific Heat Capacity at Constant Volume Unit Conversion ↻



- η_a Thermal Efficiency of Atkinson Cycle
- η_d Efficiency of Diesel Cycle
- η_e Thermal Efficiency of Ericsson Cycle
- η_i Indicated Thermal Efficiency
- η_l Thermal Efficiency of Lenoir Cycle
- η_o Efficiency of Otto Cycle
- η_r Relative Efficiency
- η_s Thermal Efficiency of Stirling Cycle
- η_{th} Thermal Efficiency of Diesel Cycle
- Φ Relative Air Fuel Ratio



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