

# Important Pauling's Electronegativity Formulas PDF



Formulas  
Examples  
with Units

## List of 11 Important Pauling's Electronegativity Formulas

### 1) Covalent Ionic Resonance Energy using Pauling's Electronegativity Formula [🔗](#)

Formula

$$\Delta_p = X_p^2$$

Example with Units

$$52.4176J = 7.24J^2$$

Evaluate Formula [🔗](#)

### 2) Covalent Radius given Pauling's Electronegativity Formula [🔗](#)

Formula

$$r_{\text{covalent}} = \sqrt{\frac{0.359 \cdot Z}{X_p - 0.744}}$$

Example with Units

$$1.1754A = \sqrt{\frac{0.359 \cdot 25}{7.24J - 0.744}}$$

Evaluate Formula [🔗](#)

### 3) Effective Nuclear Charge given Pauling's Electronegativity Formula [🔗](#)

Formula

$$Z = \frac{(X_p - 0.744) \cdot (r_{\text{covalent}}^2)}{0.359}$$

Example with Units

$$25.1951 = \frac{(7.24J - 0.744) \cdot (1.18A^2)}{0.359}$$

Evaluate Formula [🔗](#)

### 4) Electron Affinity of element using Pauling's Electronegativity Formula [🔗](#)

Formula

$$E.A = \left( (X_p + 0.2) \cdot \left( \frac{2}{0.336} \right) \right) - IE$$

Evaluate Formula [🔗](#)

Example with Units

$$17.0857J = \left( (7.24J + 0.2) \cdot \left( \frac{2}{0.336} \right) \right) - 27.2J$$

### 5) Ionization Energy of Element using Pauling's Electronegativity Formula [🔗](#)

Formula

$$IE = \left( (X_p + 0.2) \cdot \left( \frac{2}{0.336} \right) \right) - E.A$$

Evaluate Formula [🔗](#)

Example with Units

$$27.1857J = \left( (7.24J + 0.2) \cdot \left( \frac{2}{0.336} \right) \right) - 17.1J$$



## 6) Pauling's Electronegativity from Allred Rochow's Electronegativity Formula ↗

Formula

$$X_P = X_{A,R} + 0.744$$

Example with Units

$$7.244 \text{J} = 6.5 \text{J} + 0.744$$

Evaluate Formula ↗

## 7) Pauling's Electronegativity from Mulliken's Electronegativity Formula ↗

Formula

$$X_P = (0.336 \cdot X_M) - 0.2$$

Example with Units

$$7.192 \text{J} = (0.336 \cdot 22 \text{J}) - 0.2$$

Evaluate Formula ↗

## 8) Pauling's Electronegativity given Bond Energies Formula ↗

Formula

$$X_P = \sqrt{E_{(A-B)} - \left( \sqrt{\frac{E_{A-A}}{E_{B-B}}} \right)}$$

Example with Units

$$7.2272 \text{J} = \sqrt{75.47 \text{J} - \left( \sqrt{20 \text{J} \cdot 27 \text{J}} \right)}$$

Evaluate Formula ↗

## 9) Pauling's Electronegativity given Effective Nuclear Charge and Covalent Radius Formula ↗

Formula

$$X_P = \left( \frac{0.359 \cdot Z}{r_{\text{covalent}}^2} \right) + 0.744$$

Example with Units

$$7.1897 \text{J} = \left( \frac{0.359 \cdot 25}{1.18 \text{A}^2} \right) + 0.744$$

Evaluate Formula ↗

## 10) Pauling's Electronegativity given IE and EA Formula ↗

Formula

$$X_P = \left( \left( \frac{0.336}{0.5} \right) \cdot ( \text{IE} + \text{EA} ) \right) - 0.2$$

Evaluate Formula ↗

Example with Units

$$29.5696 \text{J} = \left( \left( \frac{0.336}{0.5} \right) \cdot ( 27.2 \text{J} + 17.1 \text{J} ) \right) - 0.2$$

## 11) Pauling's Electronegativity given Individual Electronegativities Formula ↗

Formula

$$X = |X_A - X_B|$$

Example with Units

$$0.2 \text{J} = |3.6 \text{J} - 3.8 \text{J}|$$

Evaluate Formula ↗



## Variables used in list of Pauling's Electronegativity Formulas above

- $E_{(A-B)}$  Actual Bond Energy given Electronegativity (Joule)
- $E_{A-A}$  Bond Energy of A<sub>2</sub> Molecule (Joule)
- $E_{B-B}$  Bond Energy of B<sub>2</sub> Molecule (Joule)
- $E_A$  Electron Affinity (Joule)
- $IE$  Ionization Energy (Joule)
- $r_{covalent}$  Covalent Radius (Angstrom)
- $X_p$  given Individual Electronegativities (Joule)
- $X_A$  Electronegativity of Element A (Joule)
- $X_{A.R}$  Allred-Rochow's Electronegativity (Joule)
- $X_B$  Electronegativity of Element B (Joule)
- $X_M$  Mulliken's Electronegativity (Joule)
- $X_p$  Pauling's Electronegativity given I.E and E.A (Joule)
- $X_p$  Pauling's Electronegativity (Joule)
- $Z$  Effective Nuclear Charge
- $\Delta_p$  Covalent Ionic Resonance Energy for X<sub>p</sub> (Joule)

## Constants, Functions, Measurements used in list of Pauling's Electronegativity Formulas above

- **Functions:** `abs`, `abs(Number)`  
The absolute value of a number is its distance from zero on the number line. It's always a positive value, as it represents the magnitude of a number without considering its direction.
- **Functions:** `sqrt`, `sqrt(Number)`  
A square root function is a function that takes a non-negative number as an input and returns the square root of the given input number.
- **Measurement:** **Length** in Angstrom (A)  
*Length Unit Conversion* ↗
- **Measurement:** **Energy** in Joule (J)  
*Energy Unit Conversion* ↗



## Download other Important Electronegativity PDFs

- **Important Allred Rochow's Electronegativity Formulas** ↗
- **Important Pauling's Electronegativity Formulas** ↗
- **Important Mulliken's Electronegativity Formulas** ↗

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